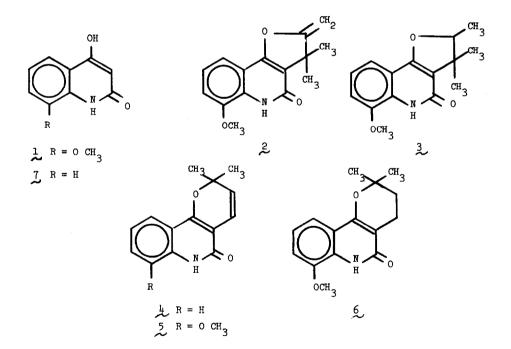
A One Step Synthesis of Flindersine¹

J. W. Huffman and T. M. Hsu Department of Chemistry and Geology Clemson University Clemson, South Carolina 29631

(Received in USA 2 December 1971; received in UK for publication 8 December 1971)

The reaction of thallous salts of nonchelated β -diketones (e.g., dimedone) with alkyl halides has been reported to afford very high yields of the corresponding 0alkylated products.² In an effort to develop a new synthesis of furanoquinoline alkaloids, we had occasion to utilize this reaction in an attempt to prepare the 4-ether of 4-hydroxy-8-methoxy-2-quinolone (1) from the corresponding thallous salt and 3-chloro-3-methyl-1-butyne. This reaction gave two products A (21%) and B (35%), both of which had the molecular formula, $C_{15}H_{15}NO_3$,³ and neither of which was the desired ether. Compound A, mp 173-175° showed the infrared and ultraviolet spectra typical of a 4-alkoxy-

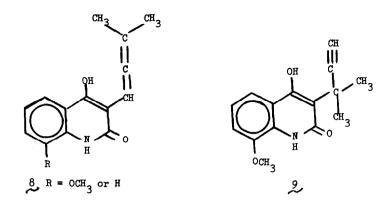


2-quinolone, however the nmr spectrum was clearly not consistent with that of the anticipated acetylenic ether and showed the presence of two vinyl protons at δ 4.35 and 4.80 with the coupling constant (3Hz) characteristic of a methylene group, and a six proton methyl singlet at δ 1.60. On the basis of these data, compound A was assigned structure 2, which was confirmed by reduction to a dihydro compound (3) mp 169-171° which showed a one proton quartet at δ 4.63 in the nmr^{3,4}.

Compound B, mp 165-167°, also lacked the spectral properties expected for the acetylenic ether, and in the nmr showed vinyl doublets at $\delta 6.78$ and 5.58 with a coupling constant of 10Hz and a six proton methyl singlet at $\delta 1.56$. These spectral data are remarkably similar to those reported for the alkaloid flindersine⁵ ($\frac{1}{4}$) and B was tentatively assigned the structure of a methoxyflindersine (5). This was confirmed by catalytic reduction to a dihydro compound ($\frac{6}{6}$)³, mp 165-167°, in which the vinyl protons had been replaced by an A_2B_2 pattern with triplets at $\delta 1.85$ and 2.67.^{3,5}

Extension of this reaction to the thallous salt of 4-hydroxy quinolone (7) gave as the principal product (28%) flindersine (4), identical to an authentic sample^{6,7}. The isomer corresponding to 2 could be detected (tlc), but was not isolated. Although there have been two previously reported syntheses of flindersine^{7,8}, the interaction of the thallous salt of 4-hydroxy-2-quinolone and 3-chloro-3-methylbutyne constitutes a one step synthesis of this alkaloid and is significantly shorter than the other routes available. In addition, this seems to be a general method for the synthesis of alkaloids of this class.

Although the reaction of the thallous salts of 1 and 7 to give 5 and 4 respectively may proceed <u>via</u> the 4-ether, followed by anti-Markovnikoff cyclization, a more plausible mechanism involves C-alkylation by way of an S_N^2 ' reaction to give 8, followed by cyclization to the angular pyranoquinoline. Furanoquinoline 2 is almost certainly formed by C-alkylation to 9 followed by cyclization, and was also obtained as the only isolable product from the reaction of 1 with 3-chloro-3-methyl-1-butyne and potassium carbonate in dimethyl formamide.



REFERENCES

- Previous paper in this series: J. W. Huffman and J. H. Cecil, <u>J. Org. Chem.</u>, <u>34</u>, 2138 (1969).
- 2. G. H. Hawks, Ph.D. dissertation, Princeton University, 1968, pages 144-162.
- 3. Satisfactory analytical data were obtained for all new compounds, and all compounds were characterized by infrared, ultraviolet and nmr spectroscopy and thin layer chromatography.
- T. R. Chamberlin and M. F. Grundon, J. <u>Chem. Soc.</u> (C), 910, 1971 have described a number of compounds similar to 3.
- 5. A. V. Robertson, <u>Aust. J. Chem.</u>, <u>16</u>, 451 (1963).
- S. Lavie, N. Danieli, R. Weitman and E. Glottier, <u>Tetrahedron</u>, <u>24</u>, 3011 (1968).
 We would like to thank Professor Lavie for a sample of flindersine.
- 7. R. F. C. Brown, G. K. Hughes and E. Ritchie, <u>Aust. J. Chem., 9</u>, 277 (1956).
- 8. F. Piozzi, P. Venturella and A. Bellino, Gazz. Chim. Ital., 99, 711 (1969).